1 ABSTRACT

Solanum paniculatum is a plant species widespread throughout tropical America, especially in 2 the Brazilian Savanna region. It is used in Brazil for culinary purposes and in folk medicine to 3 4 treat liver and gastric dysfunctions, as well as hangovers. Fractionation of the ethanolic extracts (70%) from aerial parts (leaves and twigs) of S. paniculatum (Solanaceae) led to the 5 isolation of the two new saponins (22R, 23S, 25R)-3 β , 6α , 23-trihydroxy-5 α -spirostane 6-O-6 β -D-xylopyranosyl-(1^{***} \rightarrow 3^{***})-O-[β -D-quinovopyranosyl(1^{***} \rightarrow 2^{**})]-O-[α -L-7 rhamnopyranosyl(1" \rightarrow 3')]-O- β -D-quinovopyranoside (1) and diosgenin 3-O- β -D-8 glucopyranosyl(1" \rightarrow 6')-O- β -D-glucopyranoside (2) together with four know compounds: 9 10 caffeic acid (3), diosgenin β -D-glucopiranoside (4), rutin (5), and quercetin 3-O- α -Lrhamnopyranosyl $(1^{"}\rightarrow 6^{"})$ -O- β -D-galactopyranoside (6). The structures of these compounds 11 were elucidated by extensive use of 1D and 2D NMR experiments along with ESI-MS 12 analyses. Different doses (31.25 - 500 mg/kg) of ethanolic extract of leaves from S. 13 paniculatum were evaluated against gastric ulcer induced by ethanol in rats. The lower dose 14 15 of extract able to promote antiulcer effect was 125 mg/kg. The treatment with S. paniculatum 16 by oral route was able to decrease gastric lesion area and also reduced levels of myeloperoxidase (MPO) in the gastric mucosa. Our results revels for the first time, steroidal 17 saponins from S. paniculatum and the antiulcer effect of this species at this lower dose. 18 19 Keywords: Solanaceae, Solanum paniculatum, steroidal saponins, antiulcer activity 20 21

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25 **1. Introduction**

The *Solanum* genus contains ca. 1500 species distributed all over the world. Plants in the genus are distributed in the tropical and subtropical regions and an estimated 1000-1100 species of the genus are found in South America regions (Willis, 1980). Due to the large number of species in the genus, the family was named Solanaceae (Kanada et al., 2012).

Many species of the genus are known for its economic importance, such as tomato (*S. lycopersicum*), egg-plant (*S. melongena*) and potato (*S. tuberosum*), and some are used in folk and traditional medicine, like as *S. americanum*, ("maria-pretinha"), used in the treatment of gastric ulcer, bladder spasm, joint pains, and as an effective vermifuge (Lorenzi, 2002).

Recents phytochemical studies of the *Solanum* species report the occurrence of flavonoids (Kanada et al., 2012), amides (Kanada et al., 2012), steroids (Kanada et al., 2012), lignans (Pinto et al., 2013), steroidal saponins (Kanada et al., 2012; Manase et al., 2012; Li et al., 2014; Chou et al., 2012; Chang et al., 2013), and steroidal alkaloids (Pinto et al., 2013; Zhang et al., 2013; Miranda et al., 2013). As part of our ongoing research on bioactive compounds from Brazilian plants for develop of the phytotherapics, we were investigated the potencial of the species *Solanum paniculatum*.

S. paniculatum, known popularly either as jurubeba, jurupeba, jubeba or juna, is a 41 neotropical weed of very common occurrence in Brazil, Paraguay, Bolivia, and Argentina, 42 used in folk medicine and for culinary purposes. This plant species is used in Brazilian folk 43 medicine as a tonic, antifever agent, bitter, and eupeptic to treat liver and gastric dysfunctions 44 and for the manufacture of beverages and culinary purposes (Miranda et al., 2013; Agra et al., 45 2007; Mesia-Vela et al., 2002). The plant is a component of various pharmaceutical 46 formulations including: syrups, infusions or decoctions, ethanolic extracts, and elixirs. Many 47 steroidal compounds have been isolated from this species, specially glycoalkaloids and 48 saponins (Siqueira & Macan, 1976; Schereiber et al., 1965; Ripperger et al., 1967a; Ripperger 49

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et al., 1967b; Ripperger et al., 1968) evaluated the antiulcerogenic effect of extract from
different part of *S. paniculatum* and the antisecretory effect was observed in fruits and root
but not in leaves (Mesia-Vela et al., 2002). Thus, the present study has been conducted to
isolated and identified some of the constituents of this species and assess the gastroprotective
effect of the ethanolic extract from aerial parts of *Solanum paniculatum*.

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56 2. Materials and methods

57 2.1. Materials and chemicals

¹H (500 MHz), ¹³C (125 MHz), and 2D NMR spectra were obtained on a Varian INOVA 11,7 58 T and with TMS as an internal reference. Electrospray ionisation (ESI) mass spectra were 59 acquired in the positive and negative ion mode on a LCQ FLEET instrument (Thermo 60 scientific) equipped with an ion-trap mass analyser. Optical rotations were measured on a 61 62 Jasco P-1020 polarimeter. Analytical HPLC was performed on a Jasco PU-2089 Plus with UV-DAD detector (model MD-2010 Plus), using a Phenomenex Luna C18(2) column (250 63 mm x 4.6 mm, 5 µm). Semipreparative HPLC was performed on a Varian with with UV-PDA 64 ProStar 330 detector, using Phenomenex Luna C18(2) (250 x 10 mm, 10 µm) column. For 65 column chromatography (CC) Sephadex LH-20 (GE-Healthcare Bio-Sciences AB, Uppsala, 66 Sweden) was used as packing material. TLC was performed using Merck silica gel 60 (>230 67 mesh) and precoated silica gel 60 PF254 plates. Spots on TLC plates were observed under UV 68 light and by spraying the plates with anisaldehyde-H₂SO₄ reagent, followed by heating at 120 69 °C. High-pressure liquid chromatography (HPLC) solvents (TFA and methanol – both \geq 70 99.9%) were HPLC grade and purchased from Tedia® (Fairfield, OH, USA). A Milli-Q 71 system (Millipore, Bedford, MA, USA) was used to prepare deionized water for all mobile 72 phases. NMR spectra were measured using $(CD_3)_2SO (\geq 99,8\%)$ and pyridine- $d_5 (\geq 99,8\%)$ as 73 solvent purchased from Sigma-AldrichTM (St. Louis, MO, USA). Other organic solvents 74

- (Ethyl acetate, methanol, and chloroform) were of analytical grade and from Synthlab[®] (São
 Paulo, Brazil).
- 77 2.2. Plant material
- 78 Aerial parts of S. paniculatum were collected in Araraquara municipality (January 2010), São
- Paulo State, Brazil. A voucher specimen (BOTU 027535, BOTU 027536, BOTU 027537) of
- 80 the plant was deposited at the Herbarium BOTU in Botucatu, São Paulo State, Brazil.
- 81 *2.3. Extraction and isolation of the compounds*
- 82 The dried and ground twigs (66.0 g) were percolated with ethanol (70%). The ethanolic
- extract (5.8 g) was lyophilized and chromatographed (2.7 g) on Sephadex LH-20 column
- 84 (67.0 x 3.0 cm), using MeOH/H₂O (7:3) as eluent. The fraction 48 (108.0 mg) was subjected
- to EFS-C18, using gradient H₂O (100%) to MeOH (100%), resulting in 19.0 mg of 1. The
- fraction 57 (350.0 mg) was chromatographed in HPLC (reverse phase C-18 250 x 10 mm,
- 87 10 μm, eluent: MeOH/H₂O (25:75) + 0,1% TFA, flow 2 mL.min⁻¹, λ = 254 nm), resulting in
- 25.0 mg of **3**. The leaves (1.0 Kg) were dried, ground and percolated with ethanol (70%). The
- ethanolic extract (282.7 g) was lyophilized and chromatographed (3.4 g) on Sephadex LH-20
- 90 column (67.0 x 3.0 cm), using MeOH/H₂O (7:3) as eluent. The fractions 32 (73.4 mg), 38
- 91 (129.0 mg), 70 (36.5 mg) and 77 (81.0 mg) yielding the compounds 2, 4, 5, and 5+6,
- 92 respectively.
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- 94 (22R, 23S, 25R)-3 β , 6 α , 23-trihydroxy-5 α -spirostane 6-O- β -D-xylopyranosyl-(1"" \rightarrow 3"')-O-
- 95 $[\beta$ -D-quinovopyranosyl(1"' \rightarrow 2')]-O- $[\alpha$ -L-rhamnopyranosyl(1" \rightarrow 3')]-O- β -D-
- 96 quinovopyranoside (1)
- 97 Yellow amorphous solid; $\left[\alpha\right]_{D}^{24}$ +90 (*c* 0.1, MeOH); ¹H and ¹³C NMR (pyridine-d₅), see Tables
- 98 1, 2, 3 and 4; ESIMS m/z (rel. int.): 1041 [M + Na]⁺ (10), 1056 [M+K]⁺ (2) and 1027 [M+Na-

- Me]⁺ (25) (calcd for C₅₀H₈₂O₂₁), others peaks *m/z* 594 [M+H-Xyl-Qui-Rha]⁺ (2), 749 [M+Na-100 Me-Xyl-Qui]⁺ (7) and 763 [M+Na-Xyl-Qui]⁺ (5); ESI-MS/MS of *m/z* 764 (70) result in *m/z*617 [M+Na-Xyl-Qui-Rha]⁺ (100).
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- 103 3-O- β -D-glucopyranosyl($1 \rightarrow 6$)-O- β -D-glucopyranoside (2)
- 104 Yellow amorphous solid; $\left[\alpha\right]_{D}^{24}$ -20 (c 0.1, MeOH); ¹H and ¹³C NMR (pyridine-d₅), see Tables
- 105 1, 2, 3 and 4; ESIMS m/z 739 $[M+H]^+$ (13), 761 $[M + Na]^+$ (40) and 777 $[M+K]^+$ (20) (calcd
- 106 for $C_{39}H_{62}O_{13}$), others peaks m/z 577 $[M+H-Gly]^+$ (4); ESI-MS/MS of m/z 762 (54) result in
- 107 m/z 599 $[M+H+Na-Gly]^+$ (44).
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- 109 *Caffeic acid (3)*
- 110 Yellow amorphous solid; ¹H-NMR (500 MHz, DMSO-d₆): δ 7.42 (d, J = 16.0 Hz, H-8), 7.04
- 111 (d, J = 1.5 Hz, H-2), 6.97 (dd, J = 1.5 and 8.0 Hz, H-5), 6.77 (d, J = 8.0 Hz, H-6) and 6.16 (d,
- 112 J = 16.0 Hz, H-7). ¹³C NMR (125 MHz, DMSO-d₆): 165.8 (C-9), 148.3 (C-3), 145.5 (C-7),
- 113 144.9 (C-4), 125.6 (C-1), 121.4 (C-6), 115.7 (C-5), 114.7 (C-2) and 114.3 (C-8). HRTOF-
- 114 ESIMS m/z 179.0364 [M-H]⁻ (calcd. C₉H₈O₄: m/z 179.0349).
- 115
- 116 Diosgenin β -D-glucopiranoside (4)
- 117 Yellow amorphous solid; $[\alpha]_D^{24}$ -12 (c 0.1, MeOH); ¹H and ¹³C NMR (pyridine-d₅), see Tables
- 118 1, 2, 3 and 4; ESIMS m/z 600 [M+H+Na]⁺ and 616 [M+H+K]⁺ (calcd for C₃₃H₄₉O₈).
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- 123 *Rutin (5)*
- Yellow amorphous solid; ¹H and ¹³C NMR (DMSO-d₆) data were compared with the information in the literature (Agrawal, 1989), and it was identified as rutin; ESIMS m/z 609 [M-H]⁻ (calcd for C₂₇H₃₀O₁₆).
- 127 *Quercetin 3-O-\alpha-L-rhamnopyranosyl (1["])* \rightarrow 6["])-*O-\beta-D-galactopyranoside (6)*
- Yellow amorphous solid; ¹H and ¹³C NMR (DMSO-d₆) data were compared with the information in the literature (Agrawal, 1989), and it was identified as quercetin 3-O- α -Lrhamnopyranosyl (1"" \rightarrow 6")-O- β -D-galactopyranoside; ESIMS *m/z* 634 [M+Na+H]⁺ (calcd for C₂₇H₃₀O₁₆).
- 132 *2.4. Gastric ulcer model induced by ethanol*

Male Wistar (200-250 g) were randomly divided into groups (n=7) received oral treatment 133 with saline (negative control), carbenoxolone 100 mg/Kg (positive control) or 70% ethanolic 134 extratct (70%) of leaves of S. paniculatum at doses of 31.25, 62.5, 125, 250 and 500 mg/Kg, 135 to obtain dose-response curve. We also included a group Sham, in which the animals were 136 subjected to the same procedure as the other groups but without the use of agents that induced 137 injury. This Sham group was used to determined only biochemical parameters (total 138 139 glutathione and myeloperoxidase) because the gastric lesion area was absent in this group. The administration was done in a fixed volume of 10 mL/Kg. After one hour after treatment, 140 1mL of ethanol 99.8% was orally administered and after one hour, all animals are killed and 141 the stomachs examined for obtaining the area (mm²) of gastric lesions using the program 142 AvSoft BioView (Robert et al., 1979). 143

144 2.5. Quantification of total glutathione (GSH)

Strips stored in the experiment of gastric ulcer induced by ethanol were weighed and stored in 146 1mL of 5% trichloroacetic acid (TCA). The total glutathione content of the stomach was 147 determined using the substance 5,5 'dithio-bis (2-nitrobenzoic acid) (DTNB) (Anderson, 148 1985). The enzymatic reaction consists of 200µL of sample protein containing 2 mg/mL, 0.2
149 M phosphate buffer (pH 8.0), 0.5 mM DTNB (2 mg in 10 mL of 1% sodium citrate) in a final
150 volume of 2 mL. The absorbance was determined at 412 nm using a spectrophotometer. The
151 concentration of total glutathione was expressed using the extinction coefficient of 13.6 mM.

152 2.6. Activity of Myeloperoxidase (MPO)

Myeloperoxidase is an enzyme found in azurophilic granules of neutrophils, serving as a 153 biochemical marker of granulocyte infiltration (Souza et al., 2004). The MPO activity is 154 proportional to the amount of neutrophils in the mucosa. Strips stored the model of gastric 155 ulcer ethanol were weighed. We used the reaction HTAB buffer (0.5% in 50 mM sodium 156 phosphate buffer, pH 6.0) that acts as lysing detergent granules of neutrophils which contains 157 the enzyme, this being released. The enzyme activity was determined by following the 158 reaction kinetics of the enzyme with hydrogen peroxide in a reaction buffer, where one unit of 159 160 MPO determined and 1 nmol/minute capable of degrading hydrogen peroxide at 25 °C (Krawisz et al., 1984). The absorbance was determined at 450 nm using a spectrophotometer. 161

162 2.7. Statistical analysis

Results were expressed as mean ± standard error of the mean (SEM) and subjected to analysis
of variance (ANOVA) with post-hoc Tukey and Dunnet tests, with 5% significance level,
using the GraphPad Prism 5 Demo.

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167 **3. Results and discussion**

The leaves and twigs of *S. paniculatum* were percolated with ethanol (70%) and each extract was fractionated over Sephadex LH-20 column. The obtained fractions from twigs were chromatographed by EFS-C18 to yield a new compound **1** and reversed-phase HPLC to yield compound **3**. The fractions from leaves yield a new compound **2**, along with the known compounds **4**, **5** and **6** (see Experimental section).

173	Compound 1, obtained as an yellow amorphous solid, showed a major ion peaks at m/z
174	$1041 [M + Na]^{+}$, $1056 [M+K]^{+}$ and $1027 [M+Na-Me]^{+}$ (calcd for $C_{50}H_{82}O_{21}$) in the positive
175	ESIMS mass spectrum. The mass spectra present others important peaks at m/z 594 [M+H-
176	Xyl-Qui-Rha] ⁺ , 749 [M+Na-Me-Xyl-Qui] ⁺ and 763 [M+Na-Xyl-Qui] ⁺ . The MS/MS of m/z
177	764 result in m/z 617 [M+Na-Xyl-Qui-146] ⁺ , corresponding to the loss of a rhamnose unit.
178	The ¹ H NMR spectrum of 1 showed signals for two tertiary methyl groups at $\delta_{\rm H}$ 0.86
179	(3H, br s, H-18) and 0.83 (3H, br s, H-19), five secondary methyl groups at $\delta_{\rm H}$ 1.52 (3H, d, J
180	= 7.5 Hz, H-27), 1.55 (3H, d, <i>J</i> = 7.0 Hz, H-21), 1.58 (3H, d, <i>J</i> = 6.0 Hz, H-6"), 1.63 (3H, d,
181	J = 6.0 Hz, H-6') and 1.70 (3H, d, $J = 6.0$ Hz, H-6"), four protons attached to the anomeric
182	carbons at $\delta_{\rm H}$ 4.76 (1H, d, J = 8.0 Hz, H-1'), 4.83 (1H, d, J = 7.5 Hz, H-1'''), 5.25 (1H, d, J =
183	7.5 Hz, H-1"") and 6.28 (1H, br d, $J = 1.5$ Hz, H-1"). The ¹³ C, DEPT 90° and 135° NMR data
184	(in pyridine-d5) showed the presence of 50 carbons divided into three quaternary, 30 methine,
185	10 methylene, and seven methyl carbons, with 11 carbon resonances being attributable to a β -
186	D-xylopyranosyl-(1 ^{""} \rightarrow 3 ^{""})- β -D-quinovopyranosyl moiety by comparing those already
187	reported sugar residues in Solanolactoside B (Lu et al., 2011) and 12 carbon resonances being
188	attributable to a α -L-rhamnopyranosyl(1" \rightarrow 3')- β -D-quinovopyranoside as listed in Tables 3
189	and 4. The resonances of the remaining 27 carbons originating the steroidal nucleous were
190	identical to (22R, 23S, 25R)-3 β , 6 α , 23-trihydroxy-5 α -spirostane, the aglycone of torvoside
191	C isolated from Solanum torvum (Yahara et al., 1996) and (22R, 23S, 25R)-3β, 6α, 23-
192	trihydroxy-5 α -espirostane 6-O- β -D-xylopyranosyl-(1 \rightarrow 3)-O- β -D-quinovopyranoside isolated
193	from <i>Solanum surattense</i> (Lu et al., 2011). The signals at $\delta_{\rm C}$ 70.9 (C-3) and 79.7 (C-6)
194	corresponding the substituent 3β -OH and carbinolic carbon, respectively.
195	The attachments of sugar chain were deduced from the ROESY-1D and gHMBC
196	experiments (Figure 2). The long range correlations between H-1' [$\delta_{\rm H}$ 4.76 (1H, d, $J = 8.0$

Hz)] of the quinovopyranosyl' unit with C-6 ($\delta_{\rm C}$ 79.7) of the aglycone, H-1'' [$\delta_{\rm H}$ 6.28 (1H, br 197 d, J = 1.5 Hz)] of rhamnopyranosyl with C-3' ($\delta_{\rm C}$ 83.6) of the quinovopyranosyl' unit, and H-198 1^{'''} [$\delta_{\rm H}$ 5.25 (1H, d, J = 7.5 Hz, H-1^{'''})] of the xylopyronosyl with C-3^{'''} ($\delta_{\rm C}$ 87.8) of the 199 quinovopyranosyl" unit were observed (see Figure 2). Those results indicated that the 200 quinovopyranosyl' unit was linked to C-6 of the aglycone, the rhamnopyranosyl unit being 201 linked to C-3' of the quinovopyranosyl' moiety, and the xylopyranosyl unit was linked to C-202 3" of the quinovopyranosyl" moiety. The linkage of quinovopyranosyl" unit at C-2 of the 203 quinovopyranosyl' was supported by the ROESY-1D correlations between $\delta_{\rm H}$ 4.76 (Qui' H-204 205 1') and $\delta_{\rm H}$ 4.07 (Qui'' H-4''), and $\delta_{\rm H}$ 4.83 (Qui'' H-1'') and $\delta_{\rm H}$ 6.28 (Rha H-1'') (Figure 2). A β -anomeric configurations of the D-quinovopyranosyl', D-quinovopyranosyl' and 206 D-xylopyranosyl moieties were indicated by large ${}^{3}J_{H-1 H-2}$ values of the coupling constants of 207 8.0, 7.5 and 7.5 Hz, respectively and ¹³C-NMR spectroscopic data (Yahara et al., 1996). The 208 α -anomeric configuration of the L-rhamnopyranosyl unit was deduced from their C-3" ($\delta_{\rm C}$ 209 71.2) and C-5" ($\delta_{\rm C}$ 70.2) resonance (Lu et al., 2011; Yahara et al., 1996). 210 On the basis of this evidence, this compound was identified as (22R, 23S, 25R)-3 β , 211 6α , 23-trihydroxy- 5α -spirostane 6-O- β -D-xylopyranosyl-(1^{""} \rightarrow 3^{""})-O-[β -D-212 quinovopyranosyl(1"' \rightarrow 2')]-O-[α -L-rhamnopyranosyl(1" \rightarrow 3')]-O- β -D-quinovopyranoside, a 213 new steroidal saponin. 214 Compound 2, obtained as an yellow amorphous solid, showed a major ion peaks at m/z215 739 $[M + H]^+$, 761 $[M + Na]^+$ and 777 $[M+K]^+$ (calcd for $C_{39}H_{62}O_{13}$) in the positive ESIMS 216 mass spectrum. The mass spectra present other important peak at m/z 577 [M+H-Glc]⁺. The 217 MS/MS of m/z 762 [M+Na+H]⁺ result in m/z 599 [M+Na+H-163]⁺, corresponding to the loss 218 219 of a glucose unit.

220	The ¹ H NMR spectrum of 2 showed signals for four methyl groups at $\delta_{\rm H}$ 0.82 (3H, br
221	s, H-18), 0.95 (3H, s, H-19), 1.13 (3H, d, <i>J</i> = 7.0 Hz, H-21) and 1.08 (3H, d, <i>J</i> = 7.0 Hz, H-27)
222	and two anomeric protons at $\delta_{\rm H}$ 4.83 (1H, br d, J = 8.0 Hz, H-1') and 5.18 (1H, d, J = 8.0 Hz,
223	H-1"). The ¹³ C, DEPT 90° and 135° NMR data (in pyridine-d5) showed the presence of 39
224	carbons divided into four quaternary, 22 methine, eight methylene, and four methyl carbons.
225	The characteristics quaternary C atoms resonance at $\delta_{\rm C}$ 109.7 (C-22) and 140.0 (C-5), indicate
226	the presence of a steroidal skeleton. The 27-methyl group in its axial position (S-configuration
227	at C-25) was deduced by the resonances of protons and carbons at C-25 ($\delta_{\rm H}$ 1.60/ $\delta_{\rm C}$ 27.5), C-
228	26 ($\delta_{\rm H}$ 3.36, 4.06/ $\delta_{\rm C}$ 65.1) and C-27 ($\delta_{\rm H}$ 1.08, d, $J = 7.0$ Hz/ $\delta_{\rm C}$ 16.3), in comparison with
229	literature data (Agrawal et al., 1995; Han et al., 1999). In the gHMBC experiment, the methyl
230	proton at $\delta_{\rm H}$ 0.95 (Me-19) showed long range correlations with the carbon at $\delta_{\rm C}$ 140.0 (C-5),
231	and the olefinic proton at $\delta_{\rm H}$ 5.41 (H-6) with the carbon at $\delta_{\rm C}$ 37.4 (C-10). This evidence
232	indicated the presence of a double bond ($\Delta^{5,6}$) in the B ring. From these results and by
233	comparison with spectral data reported in the literature (Agrawal et al., 1995; Han et al.,
234	1999), the aglycone of 2 was identified as (25 <i>S</i>)-spirost-5-ene (diosgenin).
235	Additionally, the ¹ H NMR spectrum of 2 displayed two signals for protons attached to
236	the anomeric carbons at $\delta_{\rm H}$ 4.83 (1H, br d, J = 8.0 Hz, H-1', Glc') and 5.18 (1H, d, J = 8.0 Hz,
237	H-1", Glc") which gave correlations in the gHSQC spectrum with signals at $\delta_{\rm C}$ 102.9, and
238	105.5, respectively, indicating the presence of two sugar units. The evaluation of chemical
239	shifts and spin-spin couplings obtained from the 2D NMR spectra allowed the identification
240	of two β -glucopyranosyl units. The relatively large ³ J value (8.0 Hz) of the anomeric proton
241	signals of Glc units indicated a β -anomeric orientation. The saccharidic chain linked at the C-
242	3 position of the aglycone was deduced by <i>g</i> HMBC correlations between $\delta_{\rm H}$ 4.83 (Glc' H-1')
243	and $\delta_{\rm C}$ 73.9 (C-3), and between $\delta_{\rm H}$ 5.18 (Glc'' H-1'') and $\delta_{\rm C}$ 70.1 (Glc' C-6') (see Figure 3).

Thus, the structure of **2** was elucidated as diosgenin 3-O- β -D-glucopyranosyl(1" \rightarrow 6')-O- β -D-glucopyranoside.

The 70% ethanolic extract from the leaves of S. paniculatum showed significant 246 gastroprotective effect at all doses tested, except the lower doses of 31.25 mg/Kg. It was 247 248 observed that the effect was dose-dependent, with the effective dose (62.5 mg/Kg) had significantly gastroprotection 38.6% compared to saline (p < 0.01), mean area injury of 142.6 249 mm², while the vehicle had an average of 232.4 mm² (Figure 4a). Doses of 125, 250 and 500 250 mg/Kg inhibited ulcer formation by 43.4, 67.6 and 71.8%, respectively. There was no 251 statistically significant difference between the positive control (carbenoxolone 100 mg/Kg) 252 and the doses of 250 and 500 mg/Kg from this ethanolic extract. 253

It was observed depletion in the levels of total glutathione present in samples of gastric tissue from animals treated with ethanolic extract (70%) from the leaves of *S. paniculatum* at different doses tested and no significant difference was observed in relation to vehicle (p<0.05). All test groups presents statistically differences from group treated with carbenoxolone (p < 0.001) (Figure 4b). This result indicates that leaf extract of this species not exert gastroprotection with participation of the glutathione pathway.

However the ethanolic extract (70%) from the leaves of *S. paniculatum* was able to reduces significantly levels of myeloperoxidase from gastric tissue at doses of 62,5 until 250 mg/kg (p<0.05) (Figure 4a). Doses of 62.5, 125, 250 and 500 mg/Kg decreased the MPO levels compared to vehicle (p<0.001) 50.3, 78.1, 91.3 and 91.4% respectively, and the doses of 250 and 500 mg/Kg were more effective in reducing the group carbenoxolone (p<0.01), which decreased by 71.0% compared to saline group.

The results show that the ethanolic extract (70%) from the leaves of *S. paniculatum* (EESPL) is able to protect the gastric mucosa to damage caused by absolute alcohol. The model of gastric ulcers induced by ethanol is classical to evaluate antiulcer activity of a drug to be tested. These agent lesive influences in several protective and lesive factors of mucosa,
reducing mucus barrier and the local blood flow, and generate ROS and increase levels of
MPO (Lopez et al., 1996).²⁷ Therefore, the gastroprotection effected by ethanolic extract of *S*. *paniculatum* may be due to the reduction of lesive factors and/or increased gastroprotective
factors, such as mucous, bicarbonate, prostaglandins, NO and glutathione.

Among the factors gastroprotective, the ethanolic extract of leaves from S. 274 *paniculatum* may act by increasing the levels of antioxidants cytoprotective agents which are 275 already present in the mucosa, but in small quantities, such as glutathione (Thomas, 2000; 276 Pavlick et al., 2002). However, the extract showed no activity in this way, which does not 277 discard the possibility that this species has antioxidant effect by other mechanisms, which will 278 be investigated in the future. However, when it comes to mitigating the damaging factors, 279 extract significantly reduced levels of MPO, an enzyme pro-oxidant indicator lesion, which is 280 281 released by neutrophils/aggregates mucosal lesions, showing that this species acts battling ROS front of an inflammatory response generated by a lesive agent (Guha et al., 2009). The 282 results present in this study could be add better productive value to this species. Previus study 283 realized from shown that this species only characterized the antiulcerogenic effects from the 284 roots and fruit at highest doses (1000 mg/kg) and this study characterized the antiulcer effect 285 from leaves at lower doses (from the dose of 250 mg/kg, by oral route) (Mesia-Vela et al., 286 2002). 287

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289 **4.** Conclusion

Six compounds were identified in the aerial parts (leaves and twigs) of *S. paniculatum*, including two new steroidal saponins. This study showed that aerial parts of *S. paniculatum* are rich source in steroidal saponins and flavonoids that are certainly responsible for the biological properties of this plant. The results obtained here complements the current

294	knowledge about S. paniculatum composition and confirm that this byproduct contains
295	diverse phytochemicals, among which saponins and flavonoids predominate, which in turn
296	could offer interesting potential applications in the food and pharmaceutical industries.
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417	Figure	Captions
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419 Figure 1. Chemical constituents isolated from *Solanum paniculatum*.

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421 Figure 2. Key ROESY-1D and gHMBC correlations of compound 1.

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423 **Figure 3.** Important gHMBC correlations observed for **2**.

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Figure 4. Gastroprotective effect of ethanolic extratct (70%) of leaves from *S. paniculatum* administered orally against the ethanol-induced ulcers (a) overall glutathione levels (b) and myeloperoxidase (c) present in the sample of gastric tissue. Results were expressed as mean \pm SEM, with subsequent Dunnet test (comparison with the negative control, saline) and Tukey test (comparison between the groups), with **p*<0.05; ** *p*<0.01; *** *p*<0.001.